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## Structure Reports

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## Key indicators

Single-crystal X-ray study
$T=173 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.002 \AA$
$R$ factor $=0.034$
$w R$ factor $=0.100$
Data-to-parameter ratio $=15.3$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## $N, N^{\prime}$-Bis(3,5-dimethyl-1H-pyrazol-1-ylmethyl)piperazine

The molecule of the title compound, $\mathrm{C}_{16} \mathrm{H}_{26} \mathrm{~N}_{6}$, is located on a crystallographic centre of inversion; as a result, there is just one half-molecule in the asymmetric unit. The piperazine ring adopts an ideal chair conformation. The substituents at the piperazine N atoms are in equatorial positions.

## Comment

A perspective view of the the title compound, (I), is shown in Fig. 1. Bond lengths and angles can be regarded as normal (Cambridge Structural Database, Version 5.27 plus one update; MOGUL Version 1.1; Allen, 2002). The piperazine ring adopts an ideal chair conformation. The sum of the bond angles at the piperazine N atoms ( $328.48^{\circ}$ ) clearly shows the pyramidal geometry. The dimethylpyrazolylmethyl residues are attached to the piperazine N atoms in equatorial positions. There are no significant $\mathrm{C}-\mathrm{H} \cdots \mathrm{N}$ or $\mathrm{C}-\mathrm{H} \cdots \pi$ contacts. The molecules in the crystal structure are held together by van der Waals interactions only.

(I)

## Experimental

The title compound was prepared according to the procedure described by Ratilainen et al. (1999).

## Crystal data

$$
\begin{aligned}
& \mathrm{C}_{16} \mathrm{H}_{26} \mathrm{~N}_{6} \\
& M_{r}=302.43 \\
& \text { Orthorhombic, } P b c a \\
& a=9.6944(9) \AA \\
& b=12.1761(11) \AA \\
& c=13.7141(12) \AA \\
& V=1618.8(3) \AA^{3} \\
& Z=4 \\
& D_{x}=1.241 \mathrm{Mg} \mathrm{~m}^{-3}
\end{aligned}
$$

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## organic papers

## Data collection

Stoe IPDS-II two-circle diffractometer
$\omega$ scans
Absorption correction: none
16864 measured reflections
1560 independent reflections

1319 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.044$
$\theta_{\text {max }}=25.8^{\circ}$
$h=-11 \rightarrow 11$
$k=-14 \rightarrow 14$
$l=-16 \rightarrow 16$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.035$

$$
\begin{aligned}
w= & 1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.0595 P)^{2}\right. \\
& +0.1979 P]
\end{aligned}
$$

where $P=\left(F_{\mathrm{o}}{ }^{2}+2 F_{\mathrm{c}}{ }^{2}\right) / 3$
$(\Delta / \sigma)_{\text {max }}<0.001$
$\Delta \rho_{\max }=0.20$ e $\AA^{-3}$
$\Delta \rho_{\min }=-0.19 \mathrm{e}^{-3}$

Table 1
Selected geometric parameters ( $\left(\AA,{ }^{\circ}\right.$ ).

| $\mathrm{N} 11-\mathrm{C} 13^{\mathrm{i}}$ | $1.4563(17)$ | $\mathrm{N} 11-\mathrm{C} 12$ | $1.4601(16)$ |
| :--- | ---: | :--- | ---: |
|  |  |  |  |
| $\mathrm{C} 11-\mathrm{N} 11-\mathrm{C} 13^{\mathrm{i}}$ | $113.99(10)$ | $\mathrm{C} 13^{\mathrm{i}}-\mathrm{N} 11-\mathrm{C} 12$ | $110.19(10)$ |
| $\mathrm{C} 11-\mathrm{N} 11-\mathrm{C} 12$ | $114.30(10)$ |  |  |
| $\mathrm{C} 13^{\mathrm{i}}-\mathrm{N} 11-\mathrm{C} 12-\mathrm{C} 13$ | $58.69(15)$ | $\mathrm{C} 12^{\mathrm{i}}-\mathrm{C} 13^{\mathrm{i}}-\mathrm{N} 11-\mathrm{C} 12$ | $-58.89(14)$ |
| $\mathrm{N} 11-\mathrm{C} 12-\mathrm{C} 13-\mathrm{N} 11^{\mathrm{i}}$ | $-58.61(15)$ |  |  |

Symmetry code: (i) $-x+1,-y+1,-z+1$.

H atoms were located in a difference electron-density map, but they were positioned geometrically and refined with fixed individual displacement parameters [ $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}(\mathrm{C})$ or $1.5 U_{\text {eq }}$ (methyl C)] using a riding model, with $\mathrm{C}-\mathrm{H}=0.95-0.99 \AA$. The methyl groups were allowed to rotate but not to tip.


Figure 1
Perspective view of the title compound with the atom numbering; displacement ellipsoids are drawn at the $50 \%$ probability level. Atoms labelled with the suffix a were generated by the symmetry code $(1-x$, $1-y, 1-z$ ).

Data collection: $X$-AREA (Stoe \& Cie, 2001); cell refinement: $X$-AREA; data reduction: $X$-AREA; program(s) used to solve structure: SHELXS97 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: PLATON (Spek, 2003); software used to prepare material for publication: SHELXL97 and PLATON.

## References

Allen, F. H. (2002). Acta Cryst. B58, 380-388.
Ratilainen, J., Airola, K., Fröhlich, R., Nieger, M. \& Rissanen, K. (1999). Polyhedron, 18, 2265-2273.
Sheldrick, G. M. (1990). Acta Cryst. A46, 467-473.
Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany. Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.
Stoe \& Cie (2001). $X$-AREA. Stoe \& Cie, Darmstadt, Germany.

